

ご案内

このプレゼンテーションは、
2023年7月9日に行われた

“第4回ライトタッチレーザー国際講演会”で発表された

**『世界で初めての新しい歯科Er:YAG
レーザーによる非熱的な殺菌機序』**

の研究報告です。

09:40-10:00, Sunday, July 9, 2023

On the production of H_2O_2 in water by LiteTouch Er:YAG-laser irradiation

Tatsuru Shirafuji¹, Jun-Seok Oh¹, Hideo Kambayashi², and Tetsuo Goto³

¹ Osaka Metropolitan University, ² Keyaki-Dori Dental Clinic, ³ NDC



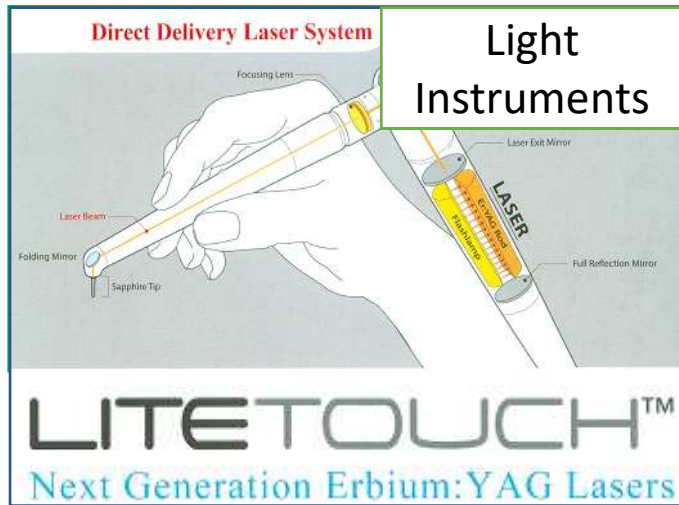
Professor Tatsuru Shirafuji specializes in plasma research at the Department of Electronics and Informatics, Faculty of Engineering, Osaka Metropolitan University, where he has published many papers. Among them, this underwater plasma theory using a dental laser is attracting attention as an opportunity to construct a new laser bactericidal theory.

Contents

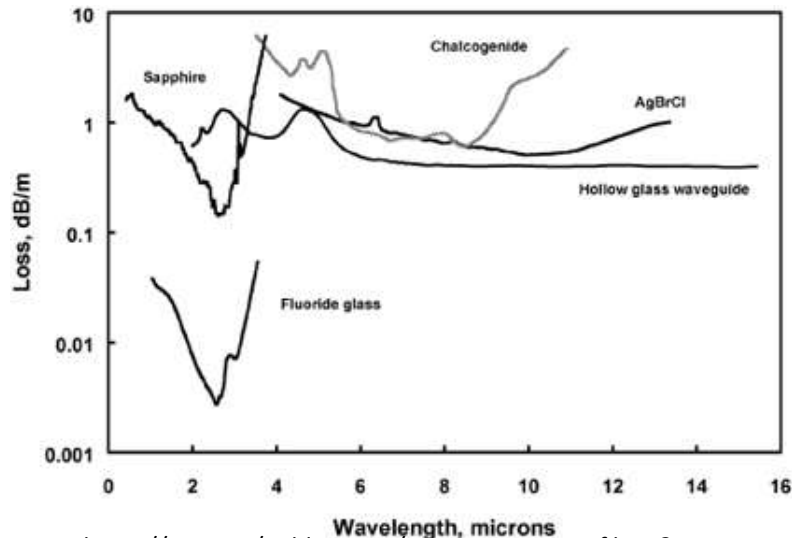
1. Reasons for selecting “LiteTouch” Er:YAG Dental Laser
2. “LITETOUCH” and “Minimally Invasive Dentistry”
3. Bactericidal effects of Er:YAG laser irradiation
4. Chemical aspects of Er:YAG laser irradiation on water
5. Analysis of Er:YAG-laser-irradiated water by **UV-Vis**
6. Discussion
7. Summary

Note: **UV-Vis**=UltraViolet-Visible Spectroscopy

Reasons for selecting “LITETOUCH” Er:YAG Dental Laser



Conventional Er:YAG lasers use optical fiber. Laser energy is fairly reduced when laser light pass through the fiber or transmission form with repeated total reflection.



https://spie.org/publications/pm135_11-13_ir_fibers?SSO=1

LITETOUCH = Fiber-less! → **Higher Power**
(L.I.H. Laser-in-Handpiece)

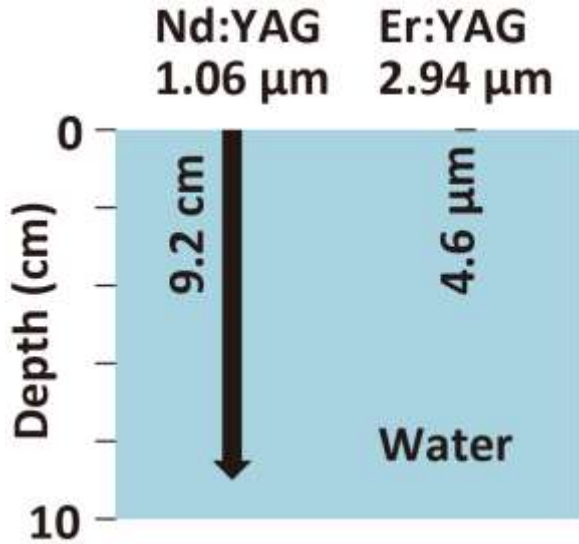
→ **High Possibility of Plasma**

Minimally Invasive Dentistry (MID) by Er:YAG Laser

No.1 “Shallow Penetration Depth”

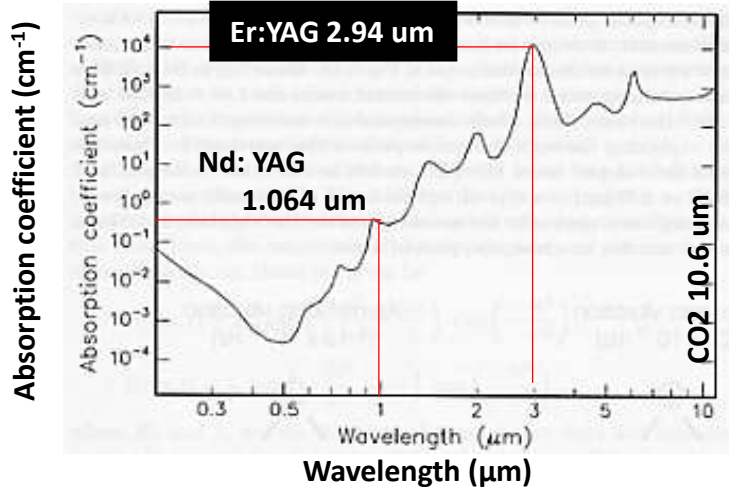
Nd:YAGは
深部の
正常組織
にまで
影響

Er:YAGは
表層部の
患部だけ
に影響



Nd:YAG $\lambda = 1.06 \mu\text{m}$, $\alpha = 1 \text{ cm}^{-1}$
Depth for $1/100 = 9.2 \text{ cm}$

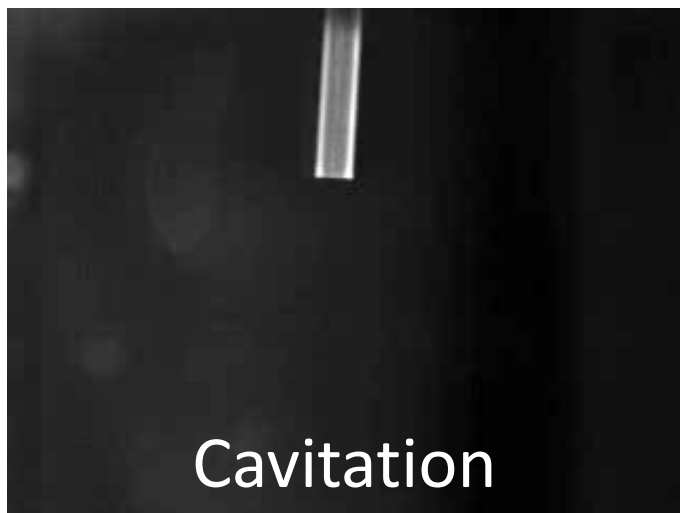
Er:YAG $\lambda = 2.94 \mu\text{m}$, $\alpha = 10^4 \text{ cm}^{-1}$
Depth for $1/100 = 4.6 \mu\text{m}$



Er:YAG is less invasive!

Minimally Invasive Dentistry (MID) by Er:YAG Laser

No.2 “Non-Thermal Dentistry”



Er:YAG is efficient! Why?

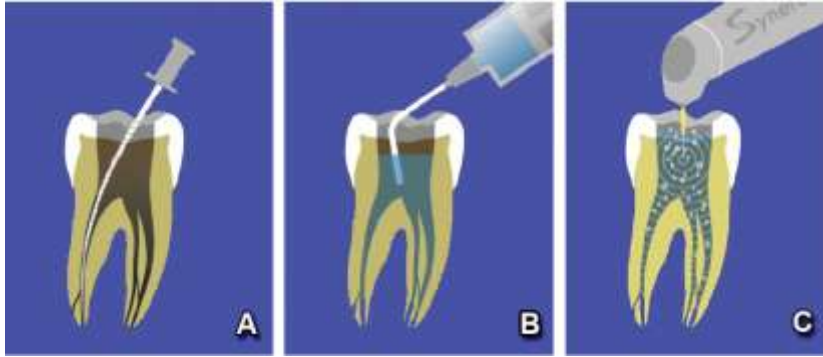
Large optical absorption coefficient (10^4 cm^{-1}) for water.

→ Laser energy can be concentrated in extremely limited areas in water.

→ Cut by Explosive Bubbles = **Non-Thermal Dentistry**

Minimally Invasive Dentistry (MID) by Er:YAG Laser

No.3 “Conventioanl Disinfection”



<https://www.dentistrytoday.com/laser-endodontic-debridement-and-canal-disinfection/>

Irrigation



L.W.M. van der Sluis, B. Verhaagen, R. Macedo, and M. Versluis:
The role of irrigation in endodontics,
in: Lasers in Endodontics,
eds. G. Olivi, R. De Moor, and E. DiVito
(Springer, 2016) p. 60 and p. 64.

Expected Effect = Root Canal Disinfection
How? = Just Irrigate it with a laser
or
“Removal” rather than “Killing”?

There are not many discussions based on a chemical point of view

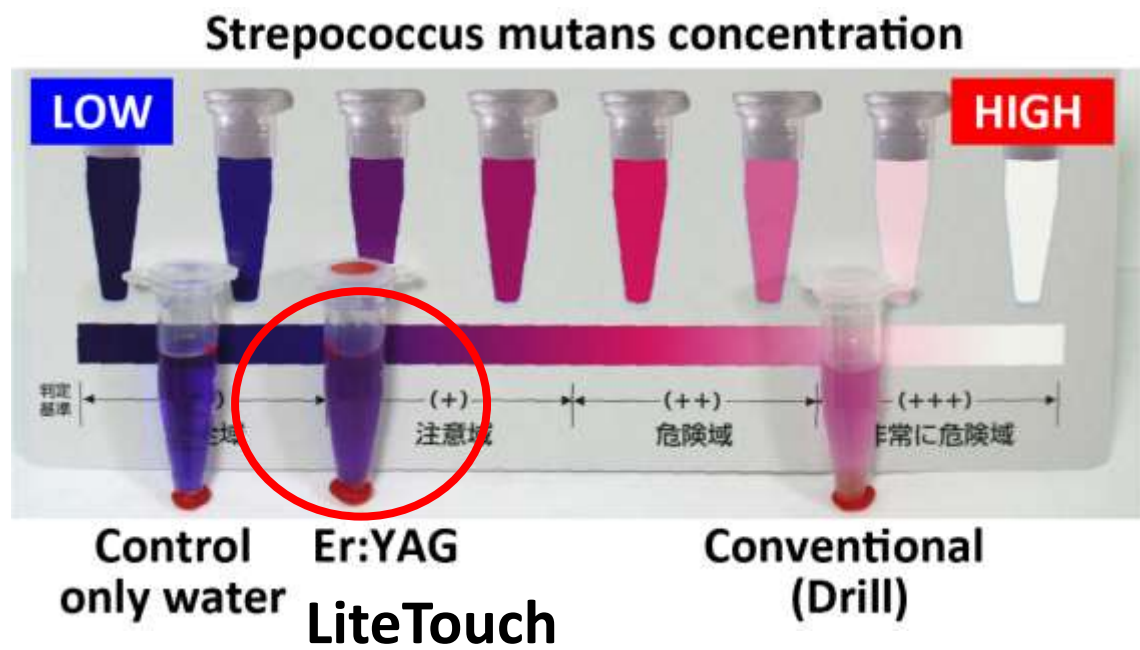
Our Previous Results

Bactericidal Effects of Er:YAG Laser Irradiation

Why?

LiteTouch Er:YAG laser irradiation on caries causes significant reduction of Streptococcus mutans concentration.

This cannot be explained by irrigation only because the target was the open surface of caries (not root canal).



J. Ueda, T. Shirafuji, H. Kambayashi, K. Takahashi, and T. Goto: On the possibility of plasma generation in water using Er:YAG laser and its application to non-invasive bactericidal, 30th Symposium on Plasma Processing (SPP-30), Actcity Hamamatsu, Seminar & Exchange Center, January 21-23, 2013, P2-36, pp. 289-290.

➔ Investigation of chemical aspects of LiteTouch Er:YAG Laser Dentistry

Chemistry of Er:YAG Laser Irradiation in or on Water

References = Sonochemistry

参考分野 = ソノケミストリー (超音波化学)

Bubble Collapse
バブル崩壊 (圧壊)



As a result H₂O is decomposed to H+OH

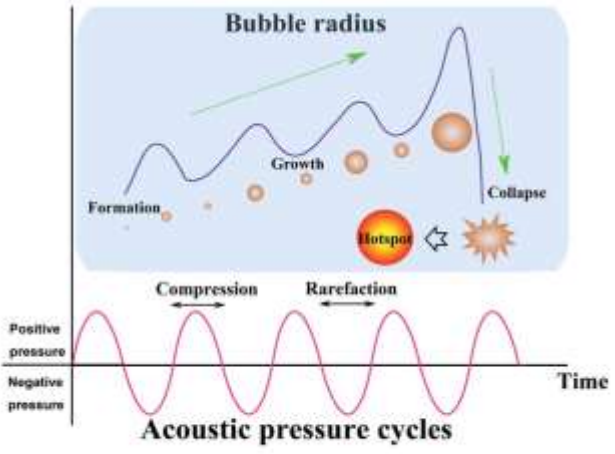


Fig. 2 Schematic illustration of the hotspot formation under acoustic cavitation.

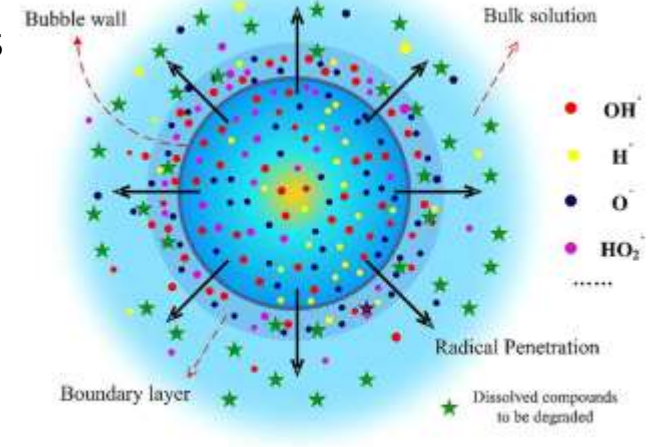
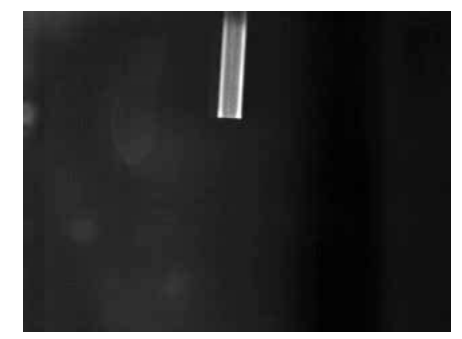
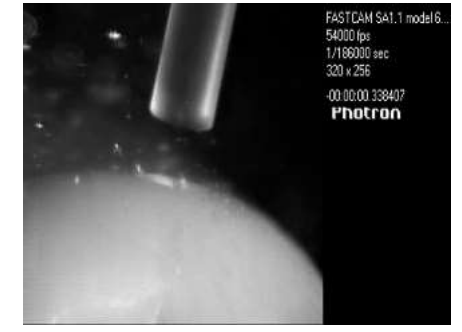


Fig. 1. Schematic of dispersion of free radicals from the cavitation bubble to the surrounding liquid.

The same phenomenon is also expected for Er:YAG

Er:YAGの場合にも同じ現象が期待される



Z. Li, J. Dong, L. Wang, Y. Zhang, T. Zhuang, H. Wang, X. Cui, and Z. Wang: A power-triggered preparation strategy of nano-structured inorganics: sonosynthesis, *Nanoscale Adv.* 3, 2423 (2021).

K. Peng, S. Tian, Y. Zhang, Q. He, and Q. Wang: Penetration of hydroxyl radicals in the aqueous phase surrounding a cavitation bubble, *Ultrasonics Sonochemistry* 91, 106235 (2022).

Experiment

UV-Vis Spectroscopy on Er:YAG-laser-irradiated Water



Quartz cuvette

With UV-Visible Spectrophotometer
Analysis of Er:YAG laser irradiated water

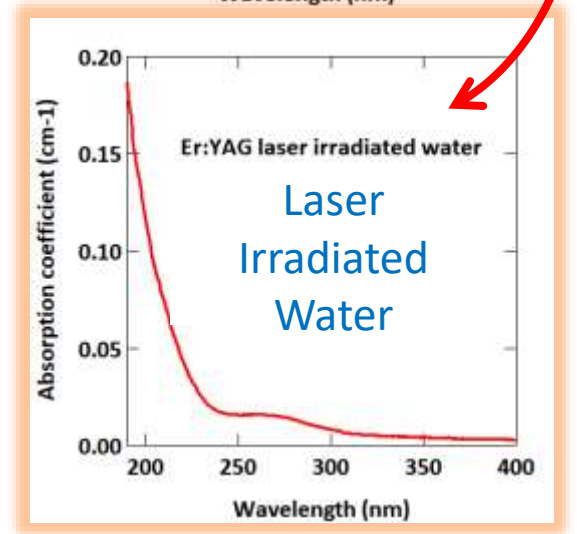
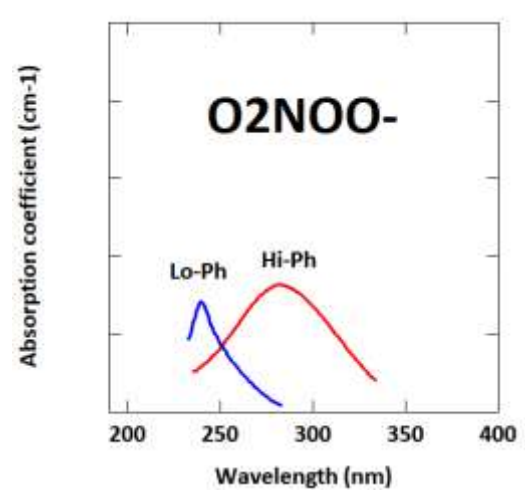
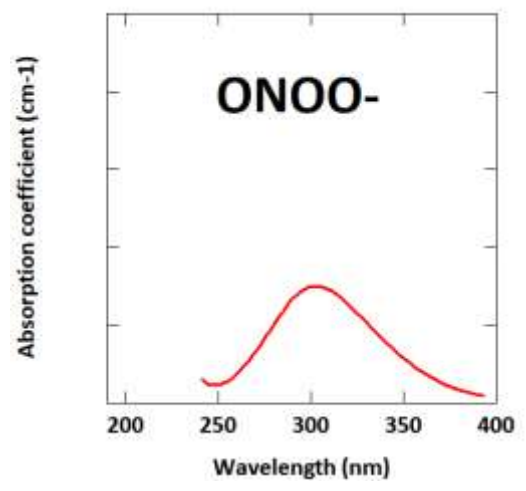
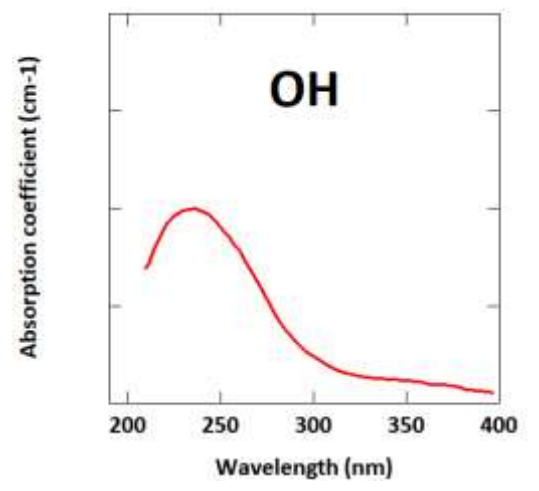
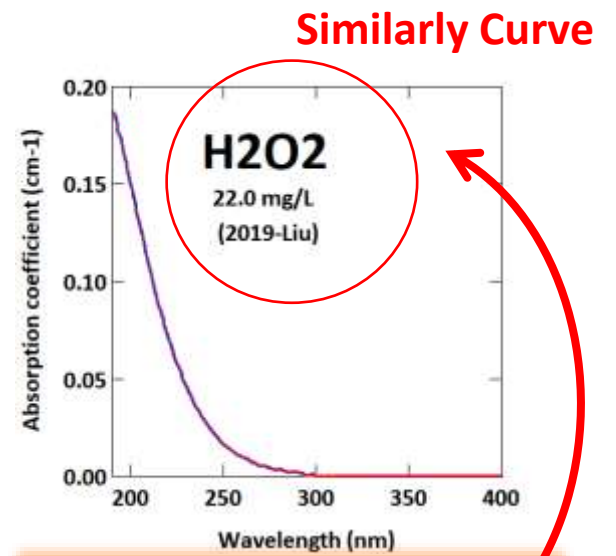
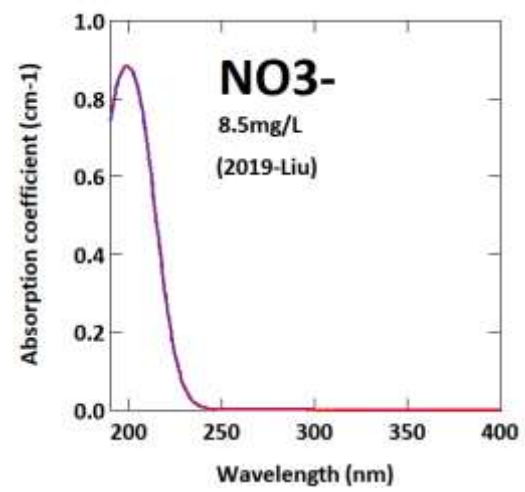
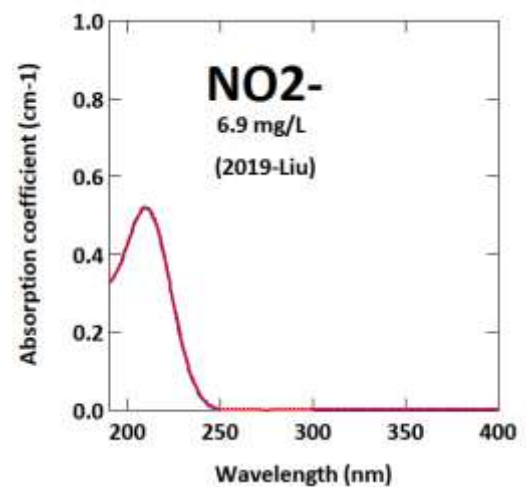
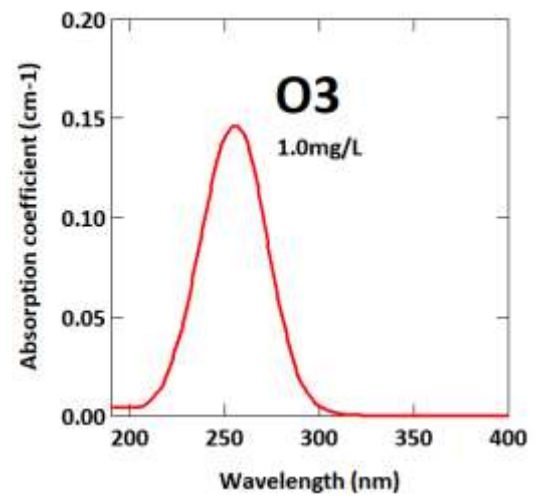
Degree of light absorption in the visible-ultraviolet region Wavelength dependence (spectrum) of wavelength



Shimadzu UV-1900
UV-Vis Spectrometer

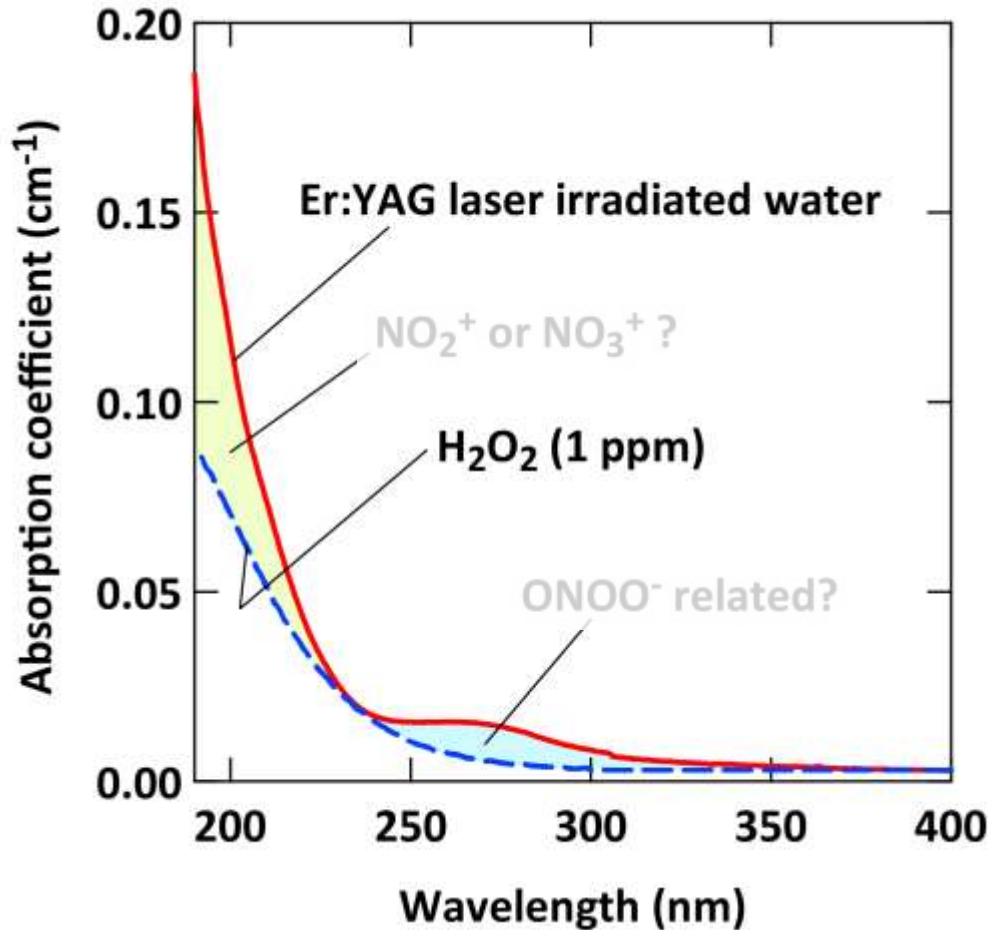
Results

Comparison of UV absorption curves of molecules



Analysis

Focusing on H2O2



The experimental result is governed by **H2O2**.

H2O2 concentration is approximately 1 ppm.

Oxidol = 3% H2O2 aqueous solution

Much less than oxidol.

However, the H2O2 concentration in the laser spot is presumed to be locally and instantaneously high.

実験結果は、**過酸化水素**で概ね説明される。

H2O2濃度はおおよそ 1 ppm.

オキシドール = 3% H2O2 水溶液

オキシドールより極めて薄い。

しかし、レーザースポットでは

局所的・瞬時的に高濃度と思われる。

Possible Mechanism of H_2O_2 Production



Cavitation Bubble

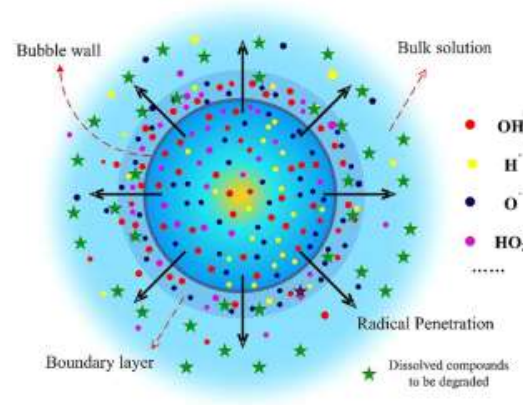
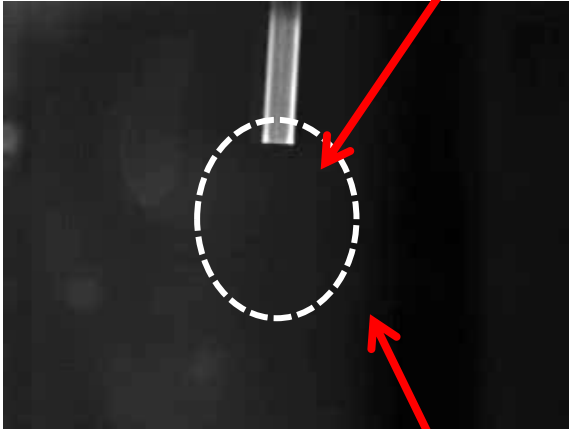
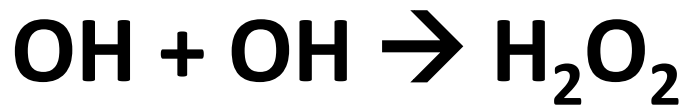


Fig. 1. Schematic of dispersion of free radicals from the cavitation bubble to the surrounding liquid.

Collapse of cavitation bubbles causes $\text{H}_2\text{O} \rightarrow \text{H} + \text{OH}$

After the collapse of the bubbles $\text{OH} + \text{OH} \rightarrow \text{H}_2\text{O}_2$

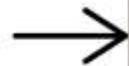


Crosscheck (Our Previous Results)

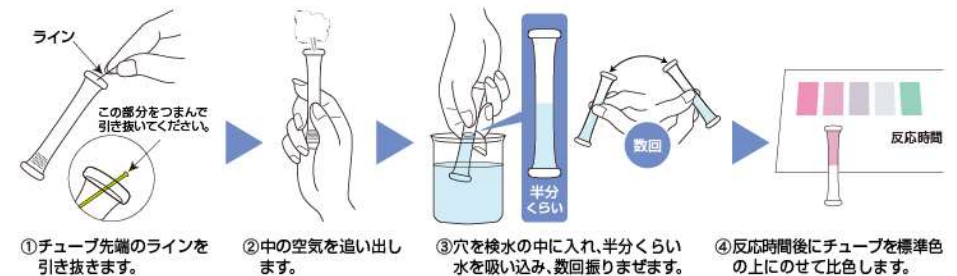
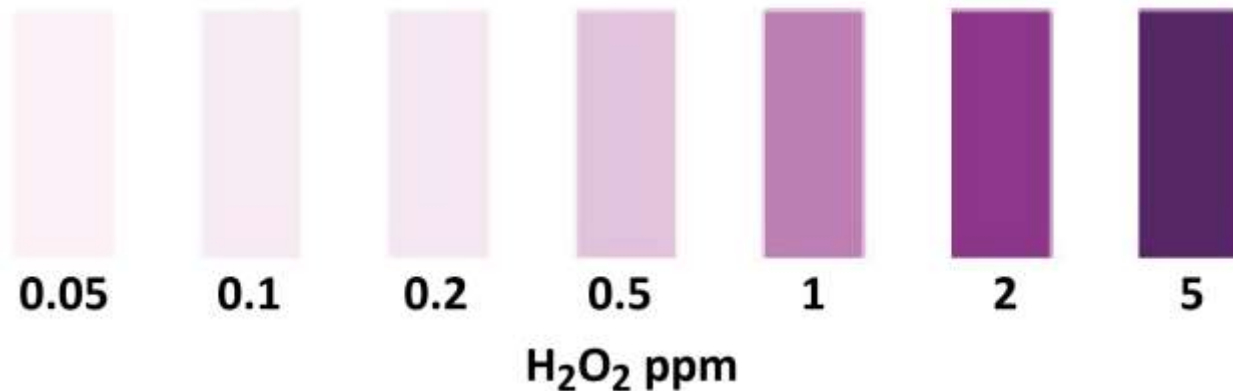
Evaluated H₂O₂ concentration does not match exactly to that deduced from UV-Vis results, but they are in the same order of magnitude.

評価された H₂O₂ 濃度は, UV-Vis の結果から推定された濃度と正確には一致しないが, 同じオーダーである.

On the day of
Er:YAG laser irradiation



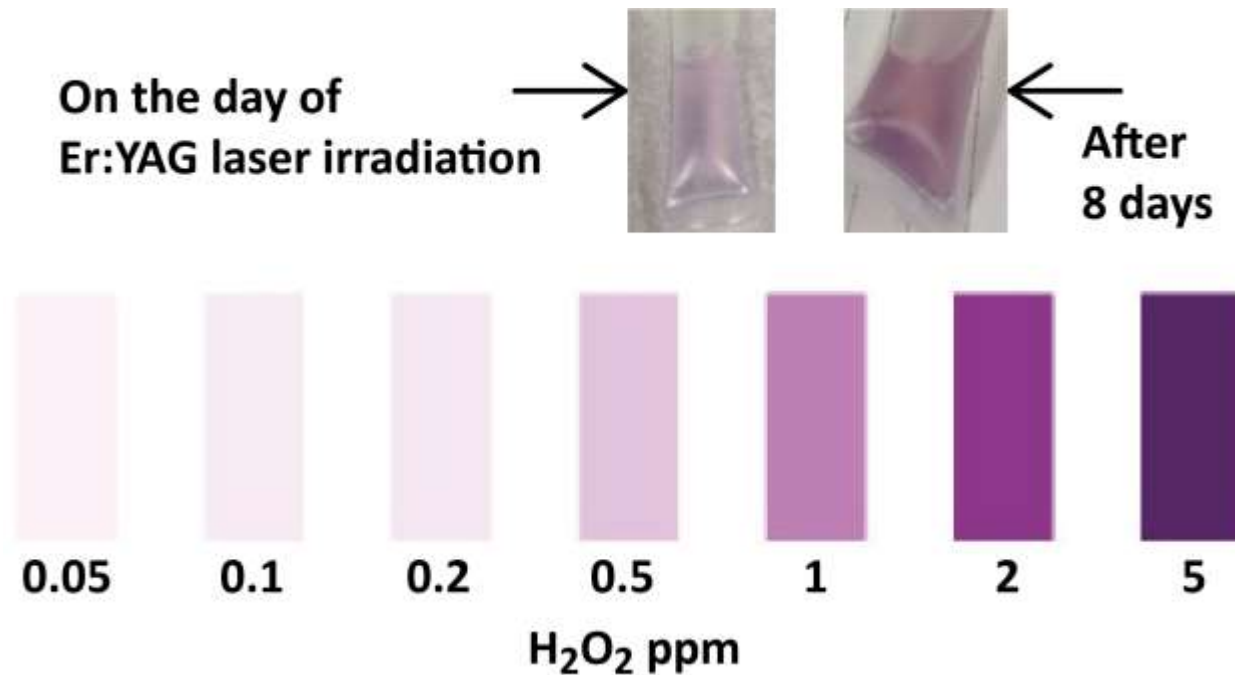
After
8 days



PACKTEST WAK-H₂O₂
(Kyoritsu Chemical-Check Lab., Corp.)

共立理化学研究所

An Interesting Result



The color of the pack test sample darkened after 8 days.

パックテストサンプルの色が8日後に濃くなった。

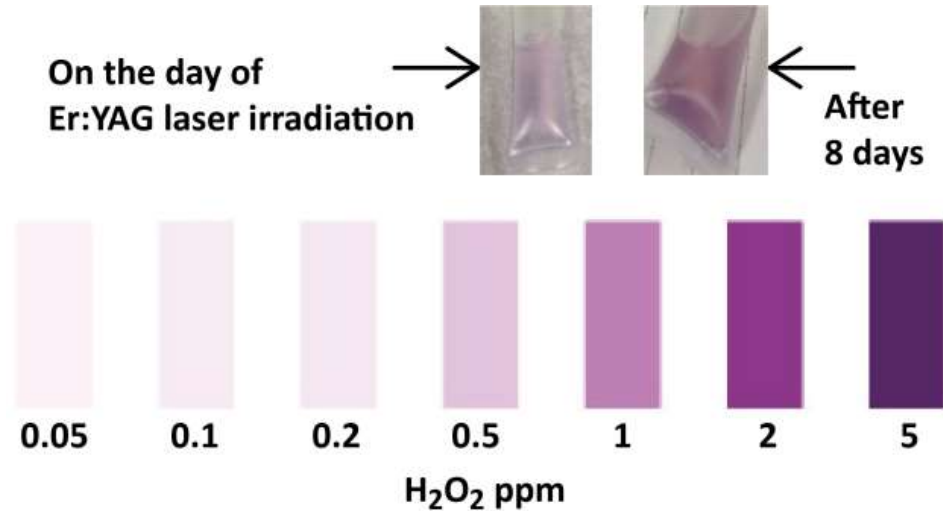
But しかし

$\text{OH} + \text{OH} \rightarrow \text{H}_2\text{O}_2$

This reaction is very fast.
Cannot explain our result.

この反応は、きわめて速い。
我々の結果を説明できない。

A Possible Mechanism



This suggests that some post-irradiation reactions continue to produce OH radicals.

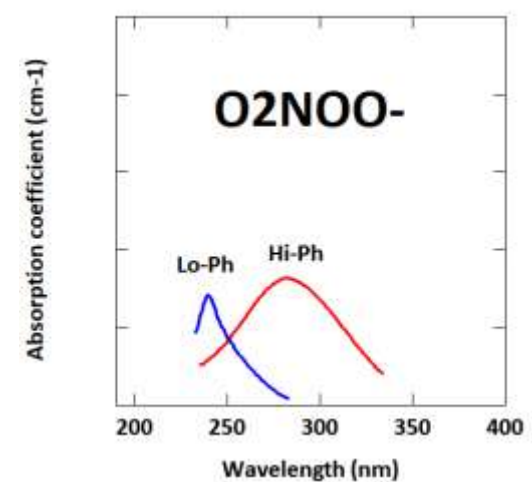
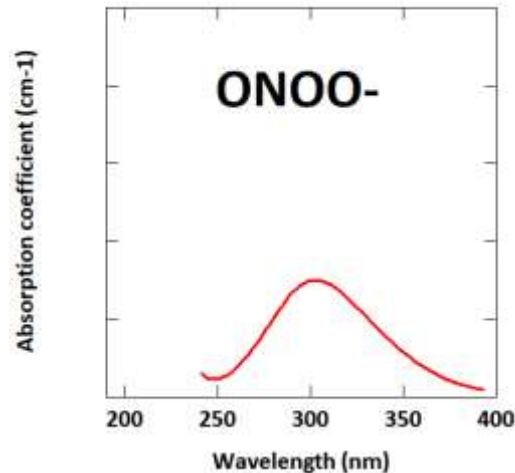
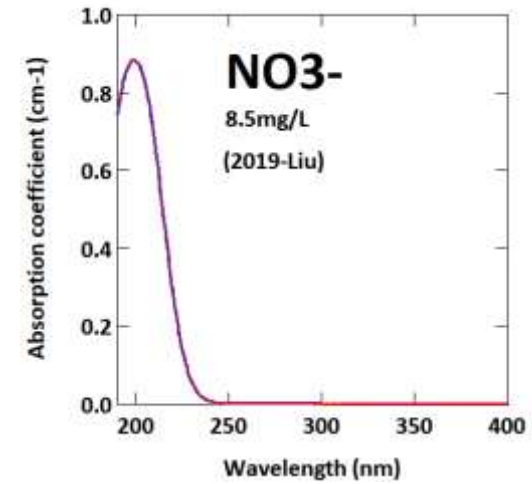
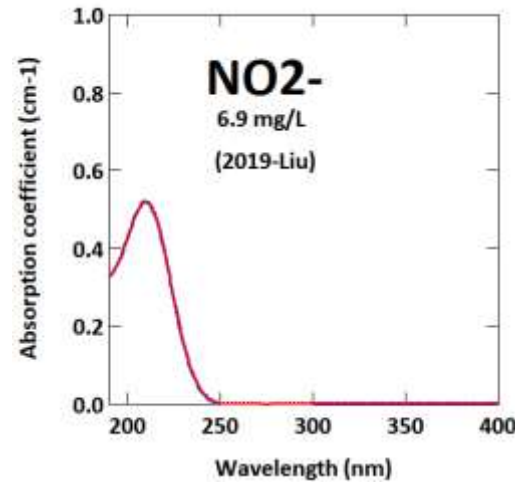
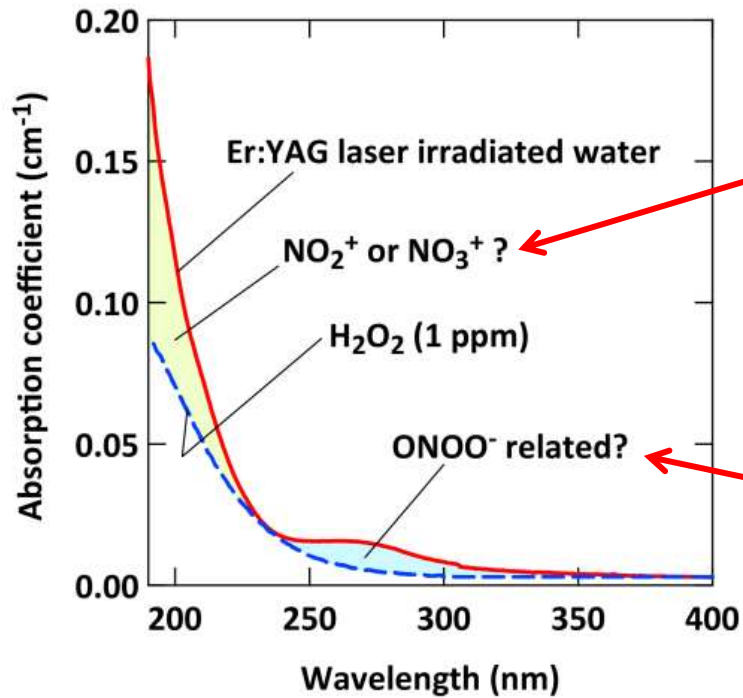
In other words, this fact is a phenomenon that demonstrates the persistence of the bactericidal action.

H₂O₂ Sustained and Increased in the PackTest for 8 days

Analysis

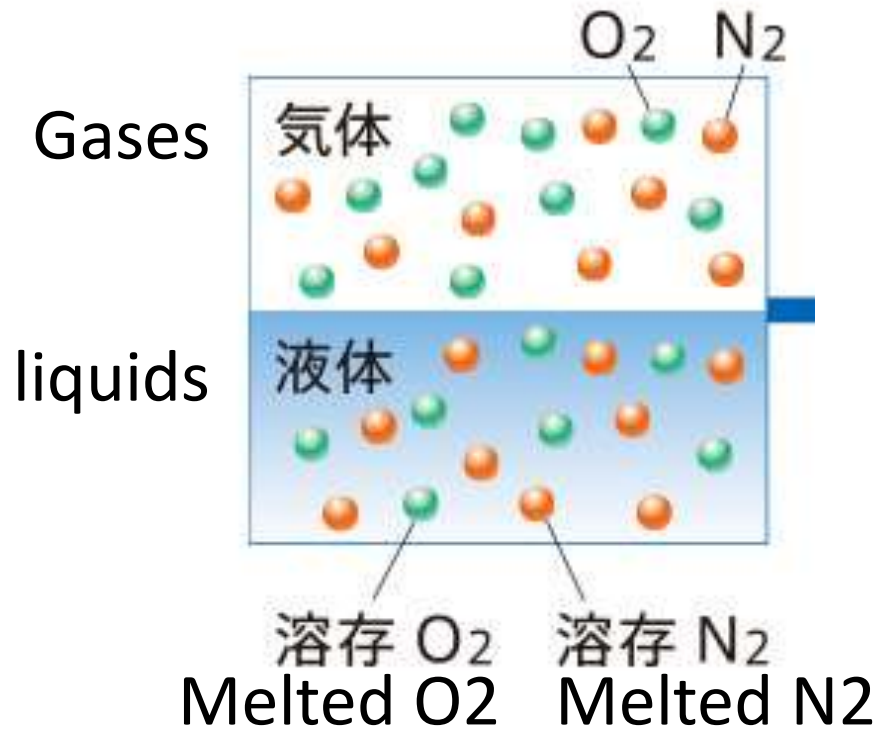
Focusing on the deviation from H2O2

The deviation from H2O2 spectra



Where does Nitrogen come from?

窒素はどこからくる？



Air is dissolved in water.

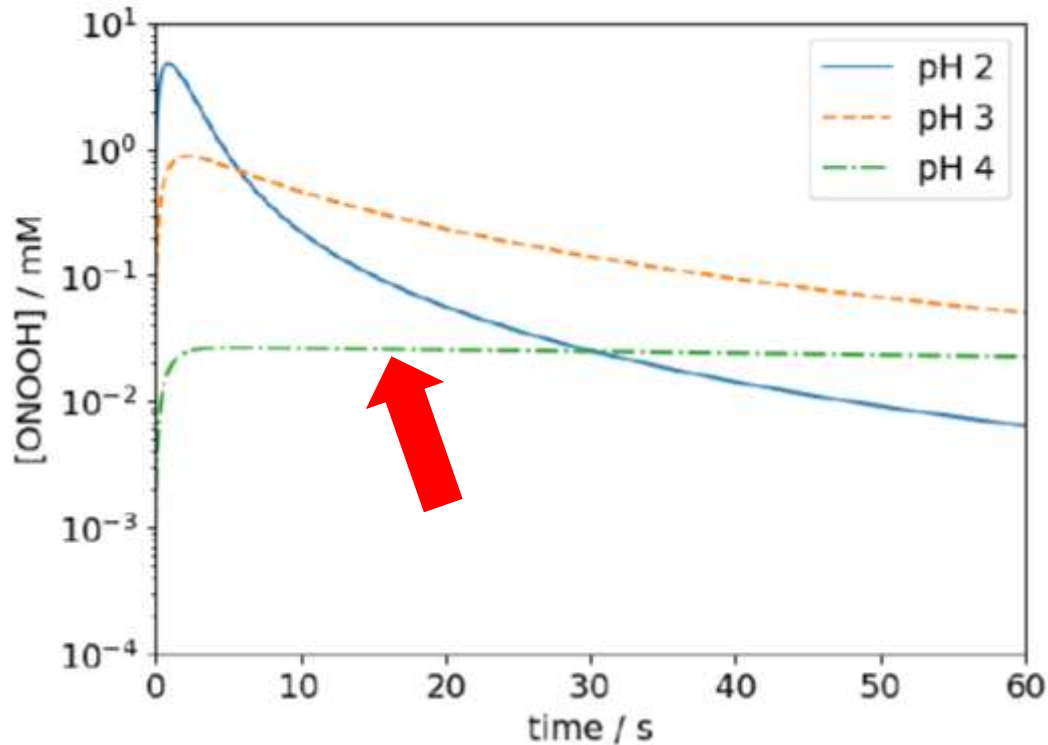
空気は水に溶けている。

Air is 20%O₂ + 80%N₂ + others.

空気は 20%O₂ + 80%N₂ + その他

A Possible Mechanism

Focusing on ONOOH/ONOO-



M. Balazinski, A. Schmidt-Bleker, J. Winter, and T. von Woedtke: Peroxynitrous acid generated in situ from acidified H₂O₂ and NaNO₂. A Suitable novel antimicrobial agent?, *Antibiotics* 10, 1003 (2021).



Equilibrium = Long-Life

平衡 = 長寿命



OH radicals are produced during gradual consumption of ONOOH

ONOOHが徐々に消費されるときに、一部がOHラジカルとなる

Summary

- 1. Er:YAG laser irradiation on water induces H₂O₂.**
- 2. This might be one of the causes of bactericidal effects.**
- 3. Our results suggest possibility of post-irradiation production of OH, which may be related to the formation of peroxyntic acid (ONOOH).**